CHEMISTRY STUDY MATERIALS FOR CLASS 12

(NCERT Based Notes of Chapter - 11) GANESH KUMAR DATE: 07/10/2020

Alcohols, Phenols and Ethers

Trihydric Alcohols: Glycerol or glycerine, CH₂OH – CH(OH)- CH₂OH is the first member of this group. Its IUPAC name is propane-I,2,3-triol.

Method of Preparation

It is obtained as a byproduct in Saponification reaction.

$$\begin{array}{c} \text{CH}_2\text{OOC}R & \text{CH}_2\text{OH} \\ \text{CHOOC}R + 3\text{NaOH} \xrightarrow{\text{Hydrolysis}} & \text{CHOH} + 3R\text{COONa} \\ \text{CH}_2\text{OOC}R & \text{CH}_2\text{OH} \\ \text{fat or oil} & \text{glycerol} \\ \end{array}$$
(where, $R = C_{17}H_{35}$ or $C_{15}H_{31}$ or $C_{17}H_{33}$)

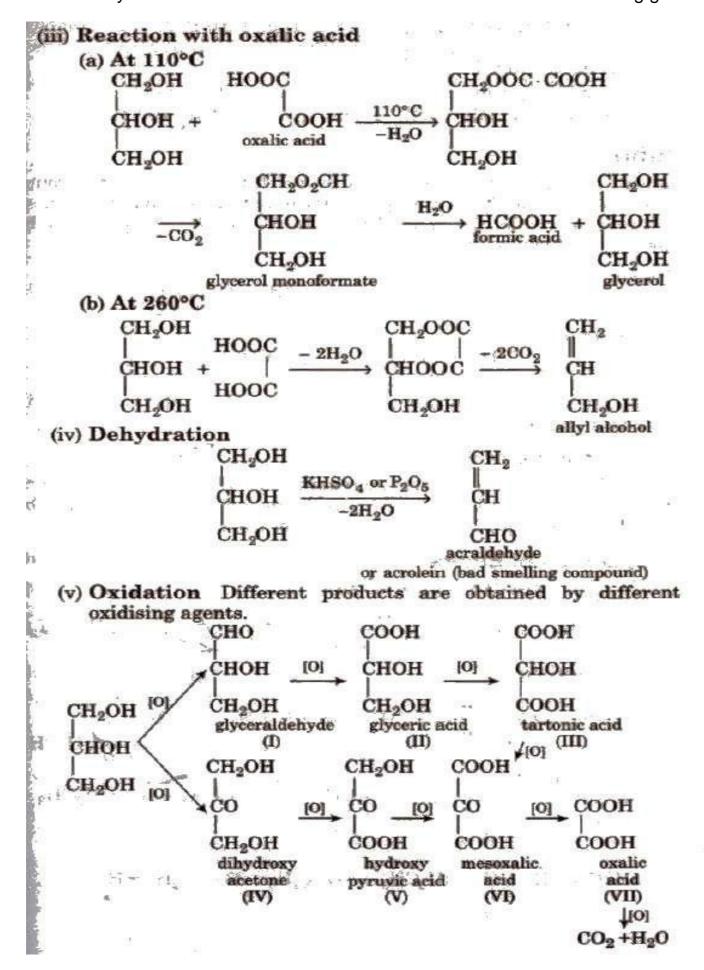
Physical Properties

- 1. It is a colourless, odourless, viscous and hygroscopic liquid.
- 2. It is sweet in taste and steam volatile.
- 3. It is soluble in water but insoluble in ether.
- 4. Due to excessive H-bonding, it is highly viscous and has high boiling point.

Chemical Properties: It gives all the general reactions given by -OR group but

2° OR is less reactive as compared to 1°. Some of its specific are:

Glyceryl trinitrate or tri nitroglycerine, when adsorbed on Kieselguhr is known as dynamite. Mixture of TNG and cellulosetrinitrate is called blasting gelatin.



Cone HNO₃ gives II; dil HNO₃ gives II and III; Bi(NO₃)₃ or NaNO₃ gives VI; Fenton's reagent or NaOBr or Br₂ water in Na₂CO₃ gives a mixture of I and IV. Solid KMnO₄ oxidizes glycerol to VII and CO₂ and H₂O. With HIO₄ (periodic acid). glycerol gives HCOOH and HCHO.
